

37. Analytical techniques

37.3 Carbon-13 NMR spectroscopy

Paper 4

Marking Scheme

Q1.

(g)	5	[1]	1
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Q2.

(d)	<table border="1"> <thead> <tr> <th>compound</th> <th>number of peaks observed</th> </tr> </thead> <tbody> <tr> <td>W</td> <td>6</td> </tr> <tr> <td>Z</td> <td>6</td> </tr> </tbody> </table>		compound	number of peaks observed	W	6	Z	6	1
	compound	number of peaks observed							
	W	6							
Z	6								
Both correct for one mark									

Q3.

(d)	14 / fourteen [1]	1
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Q4.

(b)	4, 110-160, 1, 25-50	either order [1]	2
	4, 110-160, 1, 160-185	either order [1]	


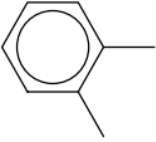
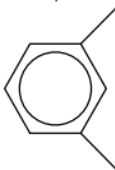
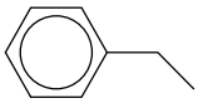
Q5.

(b)	5 5 5 4 4	2
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Q6.

(f)	6 5 [1] 4 3 [1]	2
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Q7.

(a)	three peaks	four peaks	4
			
	five peaks	six peaks	
			
correct isomers and correct assignment to peaks: mark as ✓✓✓✓✓✓			

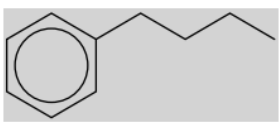
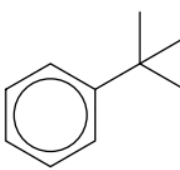
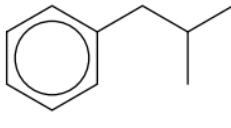
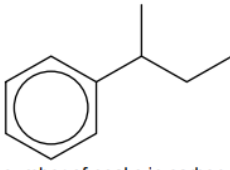
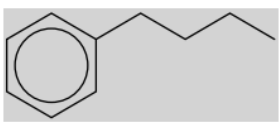
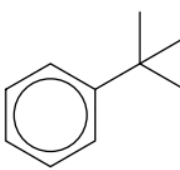
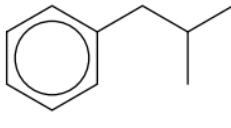
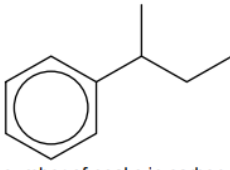
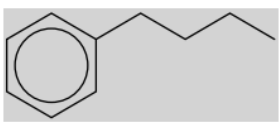
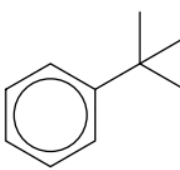
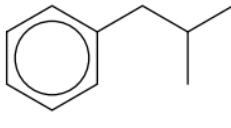
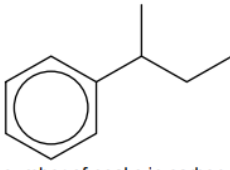
Q8.

(a)	6	[1]	1
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Q9.

(b)	benzene-1,3-dicarboxylic acid	5	[1]	2
	benzene-1,4-dicarboxylic acid	3	[1]	

Q10.

(c)(i)	$\text{CH}_3\text{CH}_2\text{CH}^+\text{CH}_3$ $(\text{CH}_3)_2\text{CHCH}_2^+$ $(\text{CH}_3)_3\text{C}^+$ Each correct structure = 1 mark	3				
(c)(ii)	<table border="1"> <tr> <td> number of peaks in carbon-13 NMR = 8</td> <td> number of peaks in carbon-13 NMR = 6</td> </tr> <tr> <td> number of peaks in carbon-13 NMR = 7</td> <td> number of peaks in carbon-13 NMR = 8</td> </tr> </table> <p>Two correct organic products = 1 mark three correct organic products = 2 marks all products linked correctly to NMR = 2 marks</p>	 number of peaks in carbon-13 NMR = 8	 number of peaks in carbon-13 NMR = 6	 number of peaks in carbon-13 NMR = 7	 number of peaks in carbon-13 NMR = 8	4
 number of peaks in carbon-13 NMR = 8	 number of peaks in carbon-13 NMR = 6					
 number of peaks in carbon-13 NMR = 7	 number of peaks in carbon-13 NMR = 8					

Q11.

(d)	<table border="1"> <thead> <tr> <th>chemical shift (δ)</th> <th>environment of the carbon atom</th> <th>hybridisation of the carbon atom</th> </tr> </thead> <tbody> <tr> <td>27</td> <td>CH_3 circled</td> <td>sp^3</td> </tr> <tr> <td>163</td> <td>COOH circled</td> <td>sp^2</td> </tr> <tr> <td>192</td> <td>$\text{C}=\text{O}(\text{COOH})$ circled</td> <td>sp^2</td> </tr> </tbody> </table> <p>Award one mark for each correct column</p>	chemical shift (δ)	environment of the carbon atom	hybridisation of the carbon atom	27	CH_3 circled	sp^3	163	COOH circled	sp^2	192	$\text{C}=\text{O}(\text{COOH})$ circled	sp^2	2
chemical shift (δ)	environment of the carbon atom	hybridisation of the carbon atom												
27	CH_3 circled	sp^3												
163	COOH circled	sp^2												
192	$\text{C}=\text{O}(\text{COOH})$ circled	sp^2												

Q12.

(e)(i)	5 peaks	1						
(e)(ii)	<table border="1"> <thead> <tr> <th>environment of carbon atom</th> <th>chemical shift range (δ)</th> </tr> </thead> <tbody> <tr> <td>carbonyl / RCOR</td> <td>190–220</td> </tr> <tr> <td>arene / benzene</td> <td>110–160</td> </tr> </tbody> </table> <p>Award one mark for each correct for each row</p>	environment of carbon atom	chemical shift range (δ)	carbonyl / RCOR	190–220	arene / benzene	110–160	2
environment of carbon atom	chemical shift range (δ)							
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